Project 5: Designing a TSA: An Overview, Video

Supplement
The Case Solutions.

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Objectives

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Design a calcium supplement that could be taken as a clear liquid

Criteria:

- It must be clear
- It has a pH between 4 and 10
- · It does not contain any toxic materials
- It has a known concentration in order to determine the amount needed for the required dose

Week 1 Procedure

- Objective: find the solubility
 properties of calcium salts using
 calcium nitrate as a source of
 soluble calcium
- Mixed Ca(NO3)2 with each of the 9 reagents provided and recorded observations
- 5 drops of a reagent mixed with 5 drops of the calcium nitrate in a single well of the well plate (*done for all 9)

Solubility Chart TheCaseSolutions.com

Reagents:	Result:
Sodium Nitrate	No precipitate; soluble
Sodium Hydrogen Phosphate	Cloudy Precipitate; insoluble
Sodium Chloride	No Precipitate; soluble
Sodium Phosphate Tribasic	Slightly Cloudy Precipitate; insoluble
Sodium Oxalate	Thick, Cloudy Precipitate; insoluble
Sodium Dihydrogen	No Precipitate; soluble
Sodium Carbonate	Cloudy Precipitate; insoluble
Sodium Acetate	No Precipitate; soluble
Sodium Sulfate	No Precipitate; soluble

Week 2

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Objective: Chemically alter calcium carbonate to make it soluble in water rather than forming a precipitate

- Used between 1-2g of Calcium nitrate and make 3 solutions of Ca+ ions by dissolving in HCl, CH3CO2H, HNO3
- Adjust pH if necessary by adding NaOH (excess acid)

Week 2 Procedure

8 CaCO₃ + 9CH₃CO₂H 8Ca+2 + 8C₂H₃O₂- +H₂O+CO₂

CaCO₃ + 2HNO₃ — Ca+2 + 2NO₃- + CO₂ + H₂O

Results

- HCl was chosen as the acid to dissolve the CaCO3 with.
- Through molar calculations, it was determined that approximately 68 mL of the solution is needed to be consumed in order to meet the daily nutritional value.

Why choose M2J2 Calcium Supplement???

- Clear, with good consistancy
- · Only have to drink a small amount
- An easy way to get recommended Calcium Intake
- DELICIOUS UNITL THE LAST DROP!!!!