Method 1- Etheno and steam

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Ethanol can be made by reacting the ethene and steam. The work and symbol equation for this is a follows.

Ethene \* Steam -> Ethanol

H4 + H2O -> C2H5OH

reading.

Blender's Credit Be Eliminated?

**Should the Ethanol** 

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Ethanol (Ethyl alcohol) is a clear and flammable oxygenated hydrocarbon. Lower concentrations of ethanol are consumed in alcoholic beverages such as whiskey, brandy and rum. Ethanol must not be drunk in high doses or will cause poisoning

Pros and con

#### Thecasesolutions.com

Pros: The ethanol is created from a renewable plant source Cons: Plants usually take a long time to grow and depend upon the weather/climate.

> Method 2-Fermentation

#### Thecasesolutions.com

Ethanol can be created through fermentation. This occurs when sugars are taken from a plant and are converted into ethanol and carbon dioxide. The symbol and word equations follow

Ward equation secasesolutions.com

Symbol equation

### Pros and cons of method 1

### Thecasesolutions.com

Pros- Ethanol can be made as long as there is a supply of ethene and

steam
Cons- Ethanol is made from
fractions of crude oil which wi

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# Should the Ethanol Blender's Credit Be Eliminated?

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## Method 1- Ethene and steam

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Ethanol can be made by reacting the ethene and steam. The word and symbol equation for this is as follows.

Ethene + Steam --> Ethanol C2H4 + H2O --> C2H5OH

Word + symbol equation

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Fthene + Steam --> Ethanol

C2H4 + H2O --> C2H5OH

## Word + symbol equation

### Thecasesolutions.com

Ethene + Steam --> Ethanol

C2H4 + H2O --> C2H5OH

## Pros and cons of method 1

## Thecasesolutions.com

Pros- Ethanol can be made as long as there is a supply of ethene and steam

Cons- Ethanol is made from fractions of crude oil which will run out eventually

### Method 2-Fermentation

### Thecasesolutions.com

Ethanol can be created through fermentation. This occurs when sugars are taken from a plant and are converted into ethanol and carbon dioxide. The symbol and word equations follow

**Word equation** 

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Sugar --> Ethanol + Carbon Dioxide **Symbol equation** 

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C66H12O6 -enzymes in yeast->

### **Word equation**

## Thecasesolutions.com

Sugar --> Ethanol + Carbon Dioxide

### **Symbol equation**

## Thecasesolutions.com

C66H12O6 -enzymes in yeast-> 2C2H5OH + 2CO2

### **Pros and cons**

### Thecasesolutions.com

Pros- The ethanol is created from a renewable plant source
Cons- Plants usually take a long time to grow and depend upon the weather/climate